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STUDY. PLAY. soluble.

Capable of being dissolved.

solution. A homogeneous

mixture of two or more

substances in a single

phase. solvent. In a

solution, the substance in

which the solute dissolves.

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suspension. Mixture in which a substance will not dissolve in another and quickly separates out if left to stand.

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Vocabulary. soluble.

solution. solvent. solute.

capable of dissolving in a

particular solvent. a

homogeneous mixture of two

or more substances uniformly

disp... in a solution, the

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substance in which the solute dissolves. in a solution, the substance that dissolves into the solvent.

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Presentation on theme:

"Chapters 12-13: Solutions Modern Chemistry."- 12 1.

Molarity, M Moles solute

Liters solution $M = \frac{M}{L} \text{ mol}$

L = Molarity is a standard

used to express the

concentration of a solution

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Moles solute Liters solution
 $M = \frac{M}{1 \text{ mol L}} = \dots$

Modern Chemistry Chapter 12 Review Answers Liquids And Solids

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Chemistry. Capillary Action.
Fluid. Surface Tension.
Vaporization. the attraction
of the surface of a liquid
to the surface of a... a
substance that can flow and
therefore take the shape of
its... a force that tends to

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pull adjacent parts of a
liquid's surfac...

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Modern Chemistry Chapter
12-Solutions Section 1-
Types of Mixtures Solutions
are homogeneous mixtures of
two or more substances in a

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single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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Modern Chemistry Chapter 12- Solutions

Title: Solutions Chapter 12
Modern Chemistry 1
SolutionsChapter 12 Modern
Chemistry Sections 1-3 Types
of Mixtures The Solution
Process Concentrations of

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Solution 1 2 Section 12.1

Types of Mixtures Chapter 12

Section 1 Types of Mixtures

p. 401-406 2 3 Vocabulary

Colloid Dispersed phase

Dispersing medium

Electrolyte Nonelectrolyte

Solute Solvent ...

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Chapter 12. Capillary
Action. Fluid. Surface
Tension. Vaporization. the
attraction of the surface of
a liquid to the surface of
a... a substance that can
flow and therefore take the
shape of its... a force that

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tends to pull adjacent parts
of a liquid's surfac...
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Chemistry Chapter 12: Solutions ...

Modern Chemistry 1 Solutions
CHAPTER 12 REVIEW Solutions
Teacher Notes and Answers
Chapter 12 SECTION 1 SHORT
ANSWER 1. c 2. a 3. b 2. a.
alcohol b. water c. the gels

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3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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CHAPTER 12 REVIEW Solutions

Modern Chemistry Chapter
12-Solutions Section 1-
Types of Mixtures Solutions
are homogeneous mixtures of
two or more substances in a
single phase.- PowerPoint

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PPT presentation Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of ...

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Chapter 12- Solutions

PowerPoint ...

Another important class 12 chemistry chapter is solutions. A solution is defined as a homogenous blend which is made by mixing two or more solutes

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and solvents. The element dissolved is called solute while the medium in which it is dissolved is called the solvent. One of the most universally used solvents is water.

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CHAPTER 12 REVIEW Solutions
- Weebly Modern Chemistry 7
Solutions CHAPTER 12 REVIEW
Solutions SECTION 3 SHORT
ANSWER Answer the following

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questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl.

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Review Answers Liquids And Solids

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